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## PREPARATION AND PROPERTIES OF AROMATIC COMPOUNDS BEARING SUBSTITUENTS WITH UNPAIRED ELECTRON

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**Abstract** Preparation of some benzoquinones, pyridiniums, and benzenoid  
 aromatics bearing substituents with unpaired electron as well as their properties,  
 especially magnetic properties, are discussed.

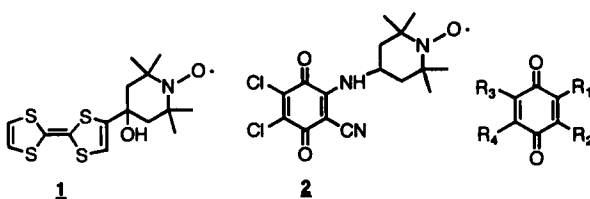
### INTRODUCTION

It is of recent interest to search for new organomagnetic materials with relevant solid state  
 properties. We are trying to prepare some organic compounds having such properties  
 and we have been interested in the preparation of three types of stable-radical-substituted  
 aromatic compounds indicated below and to arrange the unpaired electrons in their  
 column structures. In this paper, we wish to report on the preparative studies and some  
 properties of them obtained so far.

### PREPARATION AND PROPERTIES

#### TEMPO-SUBSTITUTED BENZOQUINONE DERIVATIVES

For the construction of CT complexes bearing substituent (s) with unpaired electron, we  
 have tried and succeeded to prepare hydroxy-TEMPO-substituted **1** and phenoxy-



- 3:**  $R_1=R_3=\text{NH-TEMPO}$   
 $R_2=\text{CN}, R_4=\text{Cl}$   
**4:**  $R_1=\text{NH-TEMPO}$   
 $R_2=R_3=R_4=\text{Cl}$   
**5:**  $R_1=R_3=\text{NH-TEMPO}$   
 $R_2=R_4=\text{Cl}$   
**6:**  $R_1=R_3=\text{NH-TEMPO}$   
 $R_2=R_4=\text{Br}$   
**Z:**  $R_1=\text{N}^+\text{H-TEMPO (Cl)}$   
 $R_3=R_4=\text{Cl}, R_2=\text{CN}$

substituted TTF derivatives as donor molecules and the CT complexes derived therefrom with several acceptors.<sup>1</sup> As acceptor molecule bearing substituent (s) with unpaired electron, we have prepared a series of TEMPO-substituted benzoquinone derivatives by the reaction of DDQ, chloranil or bromanil with amino-TEMPO.<sup>2</sup> It was found that mono-substituted derivative **2** was the major product in the reaction of DDQ with amino-TEMPO, whereas bis-substituted derivatives (**5**, **6**) were the major products in the reaction of chloranil or bromanil with amino-TEMPO. Compound **2** gave the salt of chloride (**7**) and **2** as well as **7** were found to give CT complexes with several donors as TTF, **1** or TMPD although others did not. Magnetic susceptibility of **2** as well as its CT complexes were measured on their polycrystalline samples by a SQUID susceptometer at temperatures between 2 K and 300 K. It was found from the susceptibility data that **2** had its susceptibility maxima around 3.8 K (Fig. 1) and the experimental data were well reproduced by theoretical data estimated by one-dimensional Heisenberg anti-ferromagnetic model with  $J = 3$  K and  $\chi_{\max} = 0.037$  emu/mol. In the CT complexes of **2**, the susceptibility data showed Curie-Weiss curve with Weiss temperature of as large as -1 K (cf. the data of TTF complex in Fig. 1) and apparent decrease of magnetic susceptibility was observed for each complex by CT formation resulting probably from the intra- and/or intermolecular singlet formation between unpaired electrons.

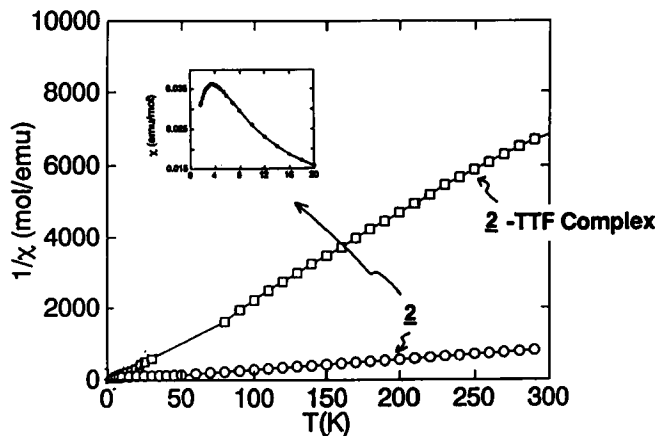
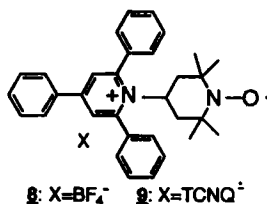


FIGURE 1 Temperature dependence of magnetic susceptibility of **2** and its TTF complex

#### TEMPO-SUBSTITUTED PYRIDINIUM DERIVATIVES

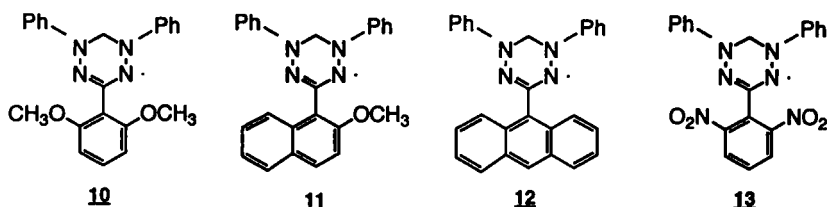


In recent papers, Awaga *et al.*<sup>3</sup> and Sugimoto *et al.*<sup>4</sup> have reported on the interesting magnetic properties in their pyridinium salts bearing substituent with unpaired electron. In the course of our studies to prepare radical salts with relevant magnetic property, we have been

interested in the preparation of N-TEMPO-substituted-pyridinium salts. Compound **8** was prepared by the reaction of corresponding pyrylium salt with amino-TEMPO and was transformed to **9** by the reaction with Li TCNQ. It was found from the susceptibility data that **8** as well as **9** showed Curie-Weiss behaviour with Weiss temperature of -1.5 K for **8** and as large as -0.2 K for **9** with no indication of parallel spin alignment between TEMPO-radicals and anion radicals of TCNQ. Further works on these and related compounds are now underway.

#### VERDAZYL-SUBSTITUTED AROMATICS

Along with the preparative studies of CT complexes or radical salts bearing substituent (s) with unpaired electron, we also have tried to prepare some radical compounds having possibly screwed conjunction between aromatic  $\pi$ -system and radical  $\pi$ -system and a series of verdazyl-substituted aromatics **10** - **13** were prepared to investigate their



magnetic behaviour in connection with their molecular as well as crystal structures.<sup>5,6</sup> It was found from the susceptibility measurements that each compound except **13** showed Curie-Weiss behavior with Weiss temperature of as large as -1 K. Compound **13** showed antiferromagnetic short range ordering (with Weiss temperature of -5 K) based possibly on one dimensionality of the system. Further studies including its X-ray crystallographic analysis are now in progress.

In summary, we have prepared three kinds of aromatic compounds bearing substituents with unpaired electron to investigate their magnetic properties, although we could see no ferromagnetic or ferrimagnetic behaviour among them yet.

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#### REFERENCES

1. S. Nakatsuji, S. Satoki, K. Suzuki, T. Enoki, N. Kinoshita, and H. Anzai, Synth. Met., **71**, 1819 (1995); S. Nakatsuji, N. Akashi, K. Suzuki, T. Enoki, N. Kinoshita, and H. Anzai, Mol. Cryst. Liq. Cryst., in press.

2. As an acceptor molecule bearing substituent with unpaired electron, Sugawara et al. have recently reported on their nitronyl nitroxide-substituted benzoquinone derivative; R. Kumai, M. M. Matsushita, A. Izuoka, and T. Sugawara, J. Am. Chem. Soc., **116**, 4523 (1994).
3. K. Awaga, T. Inabe, U. Nagashima, T. Nakamura, M. Matsumoto, Y. Kawabata, and Y. Maruyama, Chem. Lett., **1991**, 1777.
4. T. Sugimoto, M. Tsujii, E. Murahashi, H. Nakatsuji, J. Yamauchi, H. Fujita, N. Hosoi, and Y. Kai, Mol. Cryst. Liq. Cryst., **232/233**, 117 (1993).
5. Mukai et al. have recently reported on the interesting magnetic properties of their verdazyl derivatives, e. g., K. Mukai, K. Nedachi, J. B. Jamali, and N. Achiwa, Chem. Phys. Lett., **214**, 559 (1993) and their related papers. We thank to Prof. Mukai of Ehime University for his kind communication on their papers.
6. E. Dormann, H. Winter, W. Dyakonow, B. Gotschy, A. Lang, H. Naarmann, and N. Walker, Ber. Bunsenges. Phys. Chem., **96**, 922 (1992).